# organic papers

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#### **Key indicators**

Single-crystal X-ray study T = 223 KMean  $\sigma(\text{C}-\text{C}) = 0.002 \text{ Å}$  R factor = 0.044 wR factor = 0.126 Data-to-parameter ratio = 14.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# Reinvestigation of 5,12-dihydro-5,12dimethylquino[2,3-b]acridine-7,4-dione

The structure of the title compound,  $C_{22}H_{16}N_2O_2$  commonly called *trans-N,N'*-dimethylquinacridone, has been reinvestigated at 223 K [the original determination was by Zavodnik, Chetkina & Val'kova (1981). *Zh. Struckt. Khim.* **22**, 188–190; and also by Ohmasa & Susse (1976). *Naturwissenschaften*, **63**, 387–388]. The structure, in space group  $P2_1/c$ , is basically in good agreement with those reported earlier, but is determined with higher precision. The entirely planar centrosymmetric molecules are arranged with major overlap of the acridine skeleton along the stacking axis.

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## Comment

The crystal structure of the title compound, (I), was reported not only by Zavodnik et al. (1981) but also by Ohmasa & Susse (1976). The present paper provides some additional results at 223 K with higher precision. We have been conducting a series of investigations on the correlation between the color in the solid state and crystal structures of quinacridone pigments. Special attention has been focused on the effect of  $N-H \cdots O$ intermolecular hydrogen bonds on the color as well as on the stability of the compounds. For this reason, we have examined three representative quinacridone compounds with various hydrogen bonds, namely unsubstituted quinacridone with two NH groups (four hydrogen bonds per molecule; Mizuguchi et al., 2002), the monomethyl derivative with one NH group (two hydrogen bonds per molecule; Zambounis & Mizuguchi, 1996; Mizuguchi & Senju, 2002) and the dimethyl derivative with no NH group (*i.e.* the title compound; no hydrogen bonds per molecule). Compound (I) has also recently attracted attention as a doping material for electroluminescence applications (Shi & Tang, 1997). In this context, the present reinvestigation has been carried out.



Our results are basically in good agreement with those of Zavodnik *et al.* (1981). The molecule is entirely planar, as shown in Fig. 1, and belongs to the point group  $C_i$ . The molecules are stacked with major overlap of the acridine skeleton along the *a* axis. Our study has R = 0.044 for 1349 reflections with  $I > 2\sigma$ , compared with R = 0.047 for only 532 reflections

© 2003 International Union of Crystallography Printed in Great Britain – all rights reserved with  $I > 3\sigma$  in Zavodnik's study, and also with R = 0.15 for 726 non-zero reflections in Ohmasa's study. Since the measurement temperatures were different (at 223 K, in our case, but room temperature in Zavodnik's study), direct comparison of the structural parameters seems rather difficult. However, the s.u. values of the bond parameters have been improved significantly in the present investigation: the values in (I) are 0.002 Å for the bond lengths C–C, C–N and C–O, whereas the values were in the range 0.005–0.007 Å in Zavodnik's study. As for the bond angles C–C–C, C–C–N, C–N–C and C–C–O, our s.u. values are 0.1°, while a range of 0.4–0.6° was reported for Zavodnik's study.

### **Experimental**

The title compound, (I), was prepared according to a method described in the literature (Zambounis & Mizuguchi, 1996). The sample was purified by sublimation at about 623 K, using a two-zone furnace (Mizuguchi, 1981). Single crystals of (I) were then grown from a solution in dimethylformamide. The present material is quite fragile, so that cutting of the crystal using a safety razor impairs the quality of the crystal. For this reason, we used a single crystal as grown.

 $D_x = 1.444 \text{ Mg m}^{-3}$ 

Mo  $K\alpha$  radiation Cell parameters from 25

reflections

 $\theta = 14.7 - 14.9^{\circ}$ 

T = 223.2 K

Platelet, red

 $\begin{array}{l} R_{\rm int} = 0.013 \\ \theta_{\rm max} = 27.5^\circ \\ h = -6 \rightarrow 0 \end{array}$ 

 $k = 0 \rightarrow 14$ 

 $l = -18 \rightarrow 18$ 

3 standard reflections

every 150 reflections

intensity decay: 1.3%

 $\mu = 0.09 \text{ mm}^{-1}$ 

 $1.18 \times 0.17 \times 0.07~\mathrm{mm}$ 

#### Crystal data

 $\begin{array}{l} C_{22}H_{16}N_{2}O_{2}\\ M_{r}=340.38\\ \text{Monoclinic, }P2_{1}/c\\ a=4.928 (3) \text{ Å}\\ b=11.103 (3) \text{ Å}\\ c=14.462 (2) \text{ Å}\\ \beta=98.39 (2)^{\circ}\\ V=782.8 (4) \text{ Å}^{3}\\ Z=2 \end{array}$ 

#### Data collection

Rigaku AFC-7R diffractometer
$\omega$ –2 $\theta$ scans
Absorption correction: $\psi$ scan
(North et al., 1968)
$T_{\min} = 0.947, T_{\max} = 0.998$
2107 measured reflections
1797 independent reflections
1349 reflections with $F^2 > 2\sigma(F^2)$

#### Refinement

Refinement on $F^2$	H-atom parameters not refined	Obmose M $e$ Susse D (1)
$R[F^2 > 2\sigma(F^2)] = 0.044$	$w = 1/[\sigma^2(F_o^2) + \{0.05[\max(F_o^2, 0) +$	Dimiasa, M. & Susse, P. (19
$wR(F^2) = 0.126$	$2F_{2}^{2}/3^{2}$	Rigaku (1998). PROCESS-
S = 1.59	$(\Delta/\sigma)_{mm} = 0.001$	Shi, J. & Tang, C. W. (1997
1747 reflections	$\Delta \rho = 0.21 \text{ e} \text{\AA}^{-3}$	Zambounis, J. & Mizuguch
119 percenters	$\Delta \rho_{\text{max}} = 0.21 \text{ C/R}$	Zavodnik, V. E., Chetkina,
118 parameters	$\Delta \rho_{\rm min} = -0.17  {\rm e  A}$	<b>22</b> , 188–190.



## Figure 1

A view of the molecular structure of (I), showing 50% displacement ellipsoids for the non-H atoms.

# Table 1 Selected geometric parameters (Å).

O1-C7	1.234 (2)	C4-C5	1.414 (2)
N1-C5	1.380 (2)	C5-C6	1.413 (2)
N1-C9	1.389 (1)	C6-C7	1.459 (2)
N1-C11	1.461 (2)	C7-C8	1.468 (2)
C1-C2	1.376 (2)	C8-C9	1.416 (2)
C1-C6	1.405 (2)	$C8 - C10^{i}$	1.393 (2)
C2-C3	1.398 (2)	C9-C10	1.392 (2)
C3-C4	1.376 (2)		

Symmetry code: (i) -x, 1 - y, -z.

All H atoms were positioned by geometrically but were not refined. Methyl H atoms were positioned on the basis of difference Fourier maps.

Data collection: *PROCESS-AUTO* (Rigaku, 1998); cell refinement: *PROCESS-AUTO*; data reduction: *TEXSAN* (Molecular Structure Corporation, 2001); program(s) used to solve structure: *SIR*88 (Burla *et al.*, 1989); program(s) used to refine structure: *TEXSAN*; molecular graphics: *ORTEP*III (Burnett & Johnson, 1996); software used to prepare material for publication: *TEXSAN*.

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